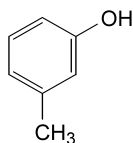


## METACRESOL

## Metacresolum



C<sub>7</sub>H<sub>8</sub>O  
[108-39-4]

*M*<sub>r</sub> 108.1

## DEFINITION

3-Methylphenol.

## CHARACTERS

*Appearance*: colourless or yellowish liquid.

*Solubility*: sparingly soluble in water, miscible with ethanol (96 per cent) and with methylene chloride.

*Relative density*: about 1.03.

mp: about 11 °C.

bp: about 202 °C.

## IDENTIFICATION

Infrared absorption spectrophotometry (2.2.24).

*Comparison*: Ph. Eur. reference spectrum of metacresol.

## TESTS

**Solution S.** Dissolve 1.5 g in carbon dioxide-free water R and dilute to 100 ml with the same solvent.

**Appearance of solution.** Freshly prepared solution S is not more opalescent than reference suspension III (2.2.1) and not more intensely coloured than reference solution BY<sub>7</sub> (2.2.2, Method II).

**Acidity.** To 25 ml of solution S add 0.15 ml of methyl red solution R. The solution is red. Not more than 0.5 ml of 0.01 M sodium hydroxide is required to change the colour of the indicator to yellow.

**Related substances.** Gas chromatography (2.2.28): use the normalisation procedure.

**Test solution.** Dissolve 1.00 g of the substance to be examined in methanol R and dilute to 100.0 ml with the same solvent.

**Reference solution (a).** Dissolve 0.10 g of cresol R, 0.10 g of *p*-cresol R and 0.10 g of the substance to be examined in methanol R and dilute to 20.0 ml with the same solvent.

**Reference solution (b).** Dilute 1.0 ml of the test solution to 100.0 ml with methanol R. Dilute 1.0 ml of this solution to 20.0 ml with methanol R.

**Column**:

– *material*: fused silica,

01/2008:2077 – *size*: *l* = 25 m, Ø = 0.25 mm,

– *stationary phase*: poly[(cyanopropyl)(methyl)][(phenyl)(methyl)]siloxane R (0.2 µm).

*Carrier gas*: helium for chromatography R.

*Flow rate*: 1.8 ml/min.

*Split ratio*: 1:30.

*Temperature*:

	Time (min)	Temperature (°C)
Column	0 - 35	100
	35 - 40	100 → 150
	40 - 50	150
Injection port		200
Detector		200

*Detection*: flame ionisation.

*Injection*: 1.0 µl.

*Relative retention* with reference to metacresol (retention time = about 28 min): impurity B = about 0.75; impurity C = about 0.98.

*System suitability*: reference solution (a):

– *resolution*: minimum 1.4 between the peaks due to impurity C and metacresol.

*Limits*:

– *impurities B, C*: for each impurity, not more than 0.5 per cent,

– *any other impurity*: for each impurity, not more than 0.1 per cent,

– *total*: not more than 1.0 per cent.

– *disregard limit*: the area of the peak due to metacresol in the chromatogram obtained with reference solution (b) (0.05 per cent).

**Residue on evaporation**: maximum 0.1 per cent.

Evaporate 2.0 g to dryness on a water-bath under a fume hood and dry at 100-105 °C for 1 h. The residue weighs a maximum of 2 mg.

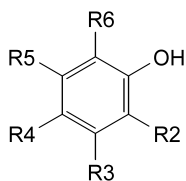
## STORAGE

In an airtight container, protected from light.

## IMPURITIES

*Specified impurities*: B, C.

*Other detectable impurities* (the following substances would, if present at a sufficient level, be detected by one or other of the tests in the monograph. They are limited by the general acceptance criterion for other/unspecified impurities and/or by the general monograph *Substances for pharmaceutical use* (2034). It is therefore not necessary to identify these impurities for demonstration of compliance. See also 5.10. *Control of impurities in substances for pharmaceutical use*): A, D, E, F, G, H, I, J, K, L, M.

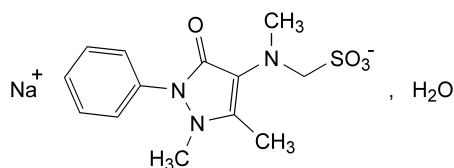


- A.  $R_2 = R_3 = R_4 = R_5 = R_6 = H$ : phenol,  
 B.  $R_2 = CH_3$ ,  $R_3 = R_4 = R_5 = R_6 = H$ : 2-methylphenol (*o*-cresol, cresol),  
 C.  $R_2 = R_3 = R_5 = R_6 = H$ ,  $R_4 = CH_3$ : 4-methylphenol (*p*-cresol),  
 D.  $R_2 = R_6 = CH_3$ ,  $R_3 = R_4 = R_5 = H$ : 2,6-dimethylphenol (2,6-xenol),  
 E.  $R_2 = C_2H_5$ ,  $R_3 = R_4 = R_5 = R_6 = H$ : 2-ethylphenol (*o*-ethylphenol),  
 F.  $R_2 = R_4 = CH_3$ ,  $R_3 = R_5 = R_6 = H$ : 2,4-dimethylphenol (2,4-xenol),  
 G.  $R_2 = R_5 = CH_3$ ,  $R_3 = R_4 = R_6 = H$ : 2,5-dimethylphenol (2,5-xenol),  
 H.  $R_2 = CH(CH_3)_2$ ,  $R_3 = R_4 = R_5 = R_6 = H$ : 2-(1-methylethyl)phenol,  
 I.  $R_2 = R_3 = CH_3$ ,  $R_4 = R_5 = R_6 = H$ : 2,3-dimethylphenol (2,3-xenol),  
 J.  $R_2 = R_4 = R_6 = H$ ,  $R_3 = R_5 = CH_3$ : 3,5-dimethylphenol (3,5-xenol),  
 K.  $R_2 = R_3 = R_5 = R_6 = H$ ,  $R_4 = C_2H_5$ : 4-ethylphenol (*p*-ethylphenol),  
 L.  $R_2 = R_5 = R_6 = H$ ,  $R_3 = R_4 = CH_3$ : 3,4-dimethylphenol (3,4-xenol),  
 M.  $R_2 = R_3 = R_5 = CH_3$ ,  $R_4 = R_6 = H$ : 2,3,5-trimethylphenol.

01/2008:1346

## METAMIZOLE SODIUM

### Metamizolum natrium



$C_{13}H_{16}N_3NaO_4S \cdot H_2O$   
 [5907-38-0]

$M_r$  351.4

#### DEFINITION

Sodium [(1,5-dimethyl-3-oxo-2-phenyl-2,3-dihydro-1*H*-pyrazol-4-yl)-*N*-methylamino]methanesulphonate monohydrate.

**Content:** 99.0 per cent to 101.0 per cent (dried substance).

#### CHARACTERS

**Appearance:** white or almost white, crystalline powder.

**Solubility:** very soluble in water, soluble in ethanol (96 per cent).

#### IDENTIFICATION

**First identification:** A, D.

**Second identification:** B, C, D.

A. Infrared absorption spectrophotometry (2.2.24).

**Comparison:** metamizole sodium CRS.

B. Dissolve 50 mg in 1 ml of *strong hydrogen peroxide solution R*. A blue colour is produced which fades rapidly and turns to intense red in a few minutes.

C. Place 0.10 g in a test tube, add some glass beads and dissolve the substance in 1.5 ml of *water R*. Add 1.5 ml of *dilute hydrochloric acid R* and place a filter paper wetted with a solution of 20 mg of *potassium iodate R* in 2 ml of *starch solution R* at the open end of the test tube. Heat gently, the evolving vapour of sulphur dioxide colours the filter paper blue. After heating gently for 1 min take a glass rod with a drop of a 10 g/l solution of *chromotropic acid, sodium salt R* in *sulphuric acid R* and place in the opening of the tube. Within 10 min, a blue-violet colour develops in the drop of the reagent.

D. 0.5 ml of solution S (see Tests) gives reaction (a) of sodium (2.3.1).

#### TESTS

**Solution S.** Dissolve 2.0 g in *carbon dioxide-free water R* and dilute to 40 ml with the same solvent.

**Appearance of solution.** Solution S is clear (2.2.1) and immediately after preparation, not more intensely coloured than reference solution BY<sub>6</sub> (2.2.2, *Method I*).

**Acidity or alkalinity.** To 5 ml of solution S, add 0.1 ml of *phenolphthalein solution R1*. The solution is colourless. Not more than 0.1 ml of 0.02 *M* sodium hydroxide is required to change the colour of the indicator to pink.

**Related substances.** Liquid chromatography (2.2.29). *Prepare the solutions immediately before use.*

**Test solution.** Dissolve 50.0 mg of the substance to be examined in *methanol R* and dilute to 10.0 ml with the same solvent.

**Reference solution (a).** Dissolve 10.0 mg of *metamizole impurity A CRS* in *methanol R* and dilute to 20.0 ml with the same solvent.

**Reference solution (b).** Dilute 1.0 ml of reference solution (a) to 20.0 ml with *methanol R*.

**Reference solution (c).** Dissolve 40 mg of *metamizole sodium CRS* in *methanol R* and dilute to 20.0 ml with the same solvent.

**Reference solution (d).** In order to prepare impurity C *in situ*, boil 10 ml of reference solution (c) under reflux for 10 min. Allow to cool to room temperature and dilute to 20.0 ml with *methanol R*.

**Reference solution (e).** To 6 ml of reference solution (a) add 1 ml of reference solution (c).

#### Column:

- *size:*  $l = 0.25$  m,  $\varnothing = 4.6$  mm;
- *stationary phase:* base-deactivated octadecylsilyl silica gel for chromatography R (5  $\mu$ m).

**Mobile phase:** mix 28 volumes of *methanol R* and 72 volumes of a buffer solution prepared as follows: mix 1000 volumes of a 6.0 g/l solution of *sodium dihydrogen phosphate R* and 1 volume of *triethylamine R*, then adjust to pH 7.0 with *strong sodium hydroxide solution R*.

**Flow rate:** 1.0 ml/min.