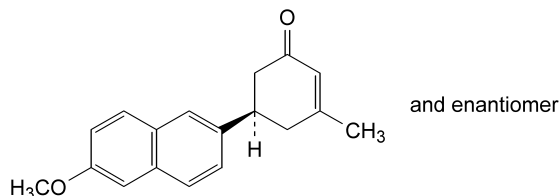
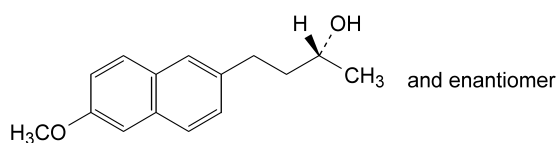


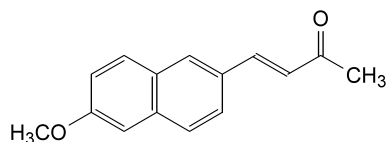
A. 3-(6-methoxynaphthalen-2-yl)-5-methylcyclohexanone,



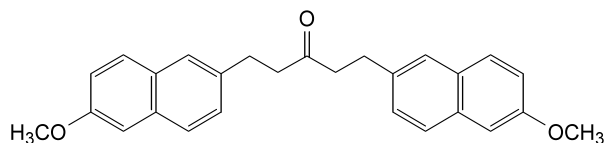
B. (5RS)-5-(6-methoxynaphthalen-2-yl)-3-methylcyclohex-2-enone,



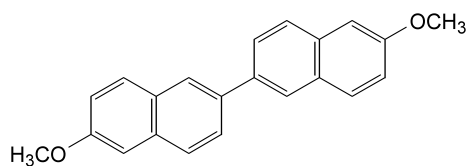
C. (2RS)-4-(6-methoxynaphthalen-2-yl)butan-2-ol,



D. (E)-4-(6-methoxynaphthalen-2-yl)but-3-en-2-one,



E. 1,5-bis(6-methoxynaphthalen-2-yl)pentan-3-one,

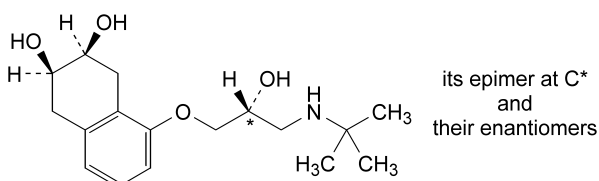


F. 6,6'-dimethoxy-2,2'-binaphthalenyl.

01/2008:1789

NADOLOL

Nadololum

its epimer at C*
and
their enantiomersC₁₇H₂₇NO₄
[42200-33-9]M_r 309.4

DEFINITION

cis-5-[3-[(1,1-Dimethylethyl)amino]-2-hydroxypropoxy]-1,2,3,4-tetrahydronaphthalene-2,3-diol.

It consists of 2 pairs of enantiomers which are present as 2 racemic compounds: racemate A and racemate B.

Content: 98.5 per cent to 101.0 per cent (dried substance).

CHARACTERS

Appearance: white or almost white, crystalline powder.

Solubility: slightly soluble in water, freely soluble in alcohol, practically insoluble in acetone.

IDENTIFICATION

Infrared absorption spectrophotometry (2.2.24).

Comparison: nadolol CRS.

TESTS

Racemate content. Infrared absorption spectrophotometry (2.2.24).

Prepare a mull in *liquid paraffin R* of the substance to be examined (dried substance), adjusting the thickness of the mull to give an absorbance reading of 0.6 ± 0.1 at 1587 cm^{-1} . Record the spectrum from 1667 to 1111 cm^{-1} , using *liquid paraffin R* as reference. Measure the absorbance A_a , corresponding to racemate A, at the maximum at 1266 cm^{-1} and the absorbance A_b , corresponding to racemate B, at the maximum at 1250 cm^{-1} . The ratio A_a/A_b is 0.72 to 1.08 (corresponding to racemate A content of between 40 per cent and 60 per cent).

Related substances. Liquid chromatography (2.2.29).

Test solution. Dissolve 0.100 g of the substance to be examined in 4.0 ml of *methanol R* and dilute to 100.0 ml with a mixture of 20 volumes of *acetonitrile R* and 80 volumes of *water R*.

Reference solution (a). Dilute 1.0 ml of the test solution to 50.0 ml with a mixture of 20 volumes of *acetonitrile R* and 80 volumes of *water R*. Dilute 5.0 ml of the solution to 50.0 ml with a mixture of 20 volumes of *acetonitrile R* and 80 volumes of *water R*.

Reference solution (b). Dissolve 20 mg of *pindolol CRS* and 20 mg of the substance to be examined in 20 ml of *methanol R* and dilute to 100.0 ml with a mixture of 20 volumes of *acetonitrile R* and 80 volumes of *water R*. Dilute 1.0 ml of the solution to 100.0 ml with a mixture of 20 volumes of *acetonitrile R* and 80 volumes of *water R*.

Reference solution (c). Dilute 5.0 ml of reference solution (a) to 20.0 ml with a mixture of 20 volumes of *acetonitrile R* and 80 volumes of *water R*.

Column:

- *size*: $l = 0.30 \text{ m}$, $\emptyset = 3.9 \text{ mm}$,
- *stationary phase*: spherical *end-capped octadecylsilyl silica gel for chromatography R* ($5 \mu\text{m}$), with a specific surface area of $700 \text{ m}^2/\text{g}$, a pore size of 0.9 nm and a carbon loading of 12 per cent.
- *temperature*: $40 \text{ }^\circ\text{C}$.

Mobile phase:

- *mobile phase A*: 5.6 g/l solution of *sodium octanesulphonate R* adjusted to pH 3.5 with a 300 g/l solution of *phosphoric acid R*,
- *mobile phase B*: *acetonitrile R*,

Time (min)	Mobile phase A (per cent V/V)	Mobile phase B (per cent V/V)
0 - 7	77	23
7 - 30	77 → 65	23 → 35
30 - 45	65	35
45 - 50	65 → 77	35 → 23
50 - 60	77	23

Flow rate: 1 ml/min.

Detection: spectrophotometer at 206 nm.

Injection: 20 µl.

Relative retention with reference to nadolol (retention time = about 18 min): impurity A = 0.25; impurity B = 0.4; impurity C (doublet) = 0.6 and 0.7; impurity D = 1.4; impurity E = 1.6; impurity F = 2.2; impurity G = 2.6.

System suitability: reference solution (b):

– resolution: minimum of 8.0 between the peaks due to nadolol and to pindolol.

Limits:

- correction factors: for the calculation of contents, multiply the peak areas of the following impurities by the corresponding correction factor: impurity A = 1.7; impurity B = 1.7; impurity C = 1.7 (multiply the sum of the areas of the 2 peaks);
- impurity A, B, C, D, E, F, G: for each impurity, not more than the area of the principal peak in the chromatogram obtained with reference solution (a) (0.2 per cent);
- any other impurity: not more than half the area of the principal peak in the chromatogram obtained with reference solution (a) (0.1 per cent);
- total: not more than 2.5 times the area of the principal peak in the chromatogram obtained with reference solution (a) (0.5 per cent);
- disregard limit: area of the principal peak in the chromatogram obtained with reference solution (c) (0.05 per cent).

Heavy metals (2.4.8): maximum 30 ppm.

1.0 g complies with limit test D. Prepare the standard using 3 ml of lead standard solution (10 ppm Pb) R.

Loss on drying (2.2.32): maximum 2.0 per cent, determined on 1.000 g by drying *in vacuo* at 60 °C for 3 h.

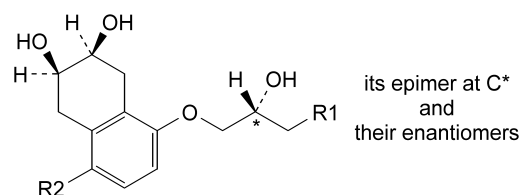
Sulphated ash (2.4.14): maximum 0.1 per cent, determined on 1.0 g.

ASSAY

Dissolve 0.250 g in 100 ml of anhydrous acetic acid R. Titrate with 0.1 M perchloric acid, determining the end-point potentiometrically (2.2.20).

1 ml of 0.1 M perchloric acid is equivalent to 30.94 mg of C₁₇H₂₇NO₄.

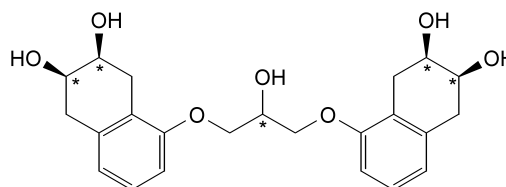
IMPURITIES



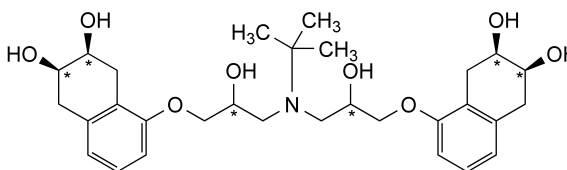
A. R1 = OH, R2 = H: *cis*-5-[(2*RS*)-2,3-dihydroxypropoxy]-1,2,3,4-tetrahydronaphthalene-2,3-diol (tetraol),

B. R1 = OCH₃, R2 = H: *cis*-5-[(2*RS*)-2-hydroxy-3-methoxypropoxy]-1,2,3,4-tetrahydronaphthalene-2,3-diol,

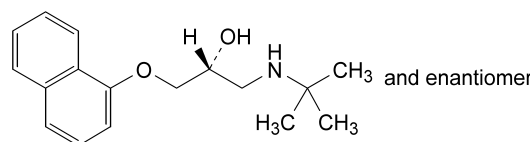
E. R1 = NH-C(CH₃)₃, R2 = I: *cis*-5-[(1,1-dimethylethylamino)-2-hydroxypropoxy]-8-iodo-1,2,3,4-tetrahydronaphthalene-2,3-diol,



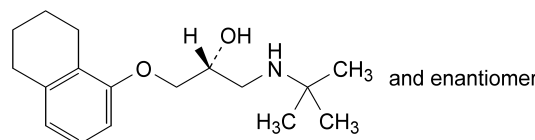
C. 5,5'-[(2-hydroxypropane-1,3-diyl)bis(oxy)]bis(*cis*-1,2,3,4-tetrahydronaphthalene-2,3-diol),



D. 5,5'-[[1,1-dimethylethylimino]bis[(2-hydroxypropane-1,3-diyl)oxy]]bis(*cis*-1,2,3,4-tetrahydronaphthalene-2,3-diol),



F. (2*RS*)-1-[(1,1-dimethylethylamino)-3-(naphthalen-1-yloxy)propan-2-ol],

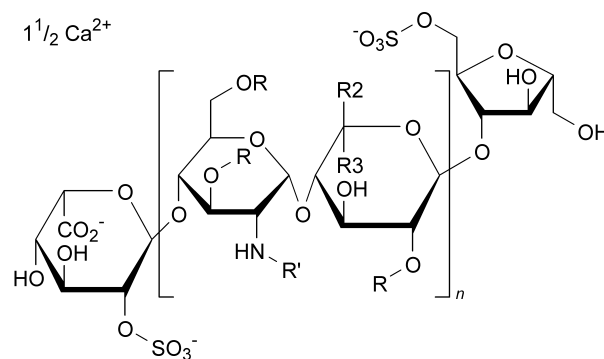


G. (2*RS*)-1-[(1,1-dimethylethylamino)-3-[(5,6,7,8-tetrahydronaphthalen-1-yl)oxy]propan-2-ol].

01/2008:1134

NADROPARIN CALCIUM

Nadroparinum calcicum



R = H or SO₃(¹/₂ Ca), R' = H or SO₃(¹/₂ Ca) or CO-CH₃
R2 = H and R3 = CO₂(¹/₂ Ca) or R2 = CO₂(¹/₂ Ca) and R3 = H