

## TESTS

**Related substances.** Liquid chromatography (2.2.29).

**Test solution.** Dissolve 0.20 g of the substance to be examined in *acetonitrile R* and dilute to 20.0 ml with the same solvent. Prepare the solution immediately before use.

**Reference solution (a).** Dissolve 5.0 mg of *triflusal impurity B CRS* in *acetonitrile R* and dilute to 10.0 ml with the same solvent.

**Reference solution (b).** Dilute 1.0 ml of reference solution (a) to 25.0 ml with *acetonitrile R*.

**Reference solution (c).** Dissolve 2.5 mg of the substance to be examined in *acetonitrile R*, add 5 ml of reference solution (a) and dilute to 10 ml with *acetonitrile R*. Prepare the solution immediately before use.

**Column:**

- size:  $l = 0.15$  m,  $\varnothing = 4.0$  mm,
- stationary phase: octadecylsilyl silica gel for chromatography *R* (4.5  $\mu$ m).

**Mobile phase:**

- mobile phase A: *acetonitrile R*,
- mobile phase B: 0.5 per cent V/V solution of phosphoric acid *R*,

Time (min)	Mobile phase A (per cent V/V)	Mobile phase B (per cent V/V)
0 - 20	20 → 70	80 → 30
20 - 25	70	30
25 - 26	70 → 20	30 → 80
26 - 30	20	80

**Flow rate:** 1.2 ml/min.

**Detection:** spectrophotometer at 237 nm.

**Injection:** 10  $\mu$ l of the test solution and reference solutions (b) and (c).

**Relative retention** with reference to triflusal (retention time = about 13 min): impurity A = about 0.3; impurity B = about 1.2; impurity C = about 1.3; impurity D = about 1.6.

**System suitability:** reference solution (c):

- resolution: minimum 3.0 between the peaks due to triflusal and impurity B.

**Limits:**

- impurity B: not more than 1.5 times the area of the corresponding peak in the chromatogram obtained with reference solution (b) (0.3 per cent),
- total of impurities other than B: not more than 0.5 times the area of the peak due to impurity B in the chromatogram obtained with reference solution (b) (0.1 per cent),
- disregard limit: 0.1 times the area of the peak due to impurity B in the chromatogram obtained with reference solution (b) (0.02 per cent).

**Heavy metals** (2.4.8): maximum 10 ppm.

Dissolve 2.0 g in 12 ml of *alcohol R* and dilute to 20 ml with *water R*. 12 ml complies with limit test B. Prepare the standard using lead standard solution (1 ppm Pb) obtained by diluting *lead standard solution (100 ppm Pb) R* with a mixture of 2 volumes of *water R* and 3 volumes of *alcohol R*.

**Loss on drying** (2.2.32): maximum 0.5 per cent, determined on 1.000 g by drying in a desiccator *in vacuo* over *diphosphorus pentoxide R*.

**Sulphated ash** (2.4.14): maximum 0.1 per cent, determined on 1.0 g in a platinum crucible.

## ASSAY

Dissolve 0.200 g in 50 ml of *ethanol R*. Titrate with 0.1 M sodium hydroxide, determining the end-point potentiometrically (2.2.20).

1 ml of 0.1 M sodium hydroxide is equivalent to 24.82 mg of  $C_{10}H_7F_3O_4$ .

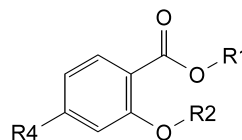
## STORAGE

In an airtight container, at a temperature not exceeding 25 °C.

## IMPURITIES

**Specified impurities:** B.

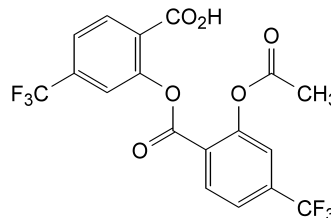
**Other detectable impurities** (the following substances would, if present at a sufficient level, be detected by one or other of the tests in the monograph. They are limited by the general acceptance criterion for other/unspecified impurities and/or by the general monograph *Substances for pharmaceutical use (2034)*. It is therefore not necessary to identify these impurities for demonstration of compliance. See also 5.10. *Control of impurities in substances for pharmaceutical use*): A, C, D.



A. R1 = H, R2 = CO-CH<sub>3</sub>, R4 = CO<sub>2</sub>H: 2-(acetyloxy)benzene-1,4-dicarboxylic acid (2-acetoxyterephthalic acid),

B. R1 = R2 = H, R4 = CF<sub>3</sub>: 2-hydroxy-4-(trifluoromethyl)benzoic acid (4-(trifluoromethyl)salicylic acid),

C. R1 = R2 = CO-CH<sub>3</sub>, R4 = CF<sub>3</sub>: acetic 2-(acetyloxy)-4-(trifluoromethyl)benzoic anhydride,



D. 2-[[2-(acetyloxy)-4-(trifluoromethyl)benzoyl]oxy]-4-(trifluoromethyl)benzoic acid.

01/2008:0868

## TRIGLYCERIDES, MEDIUM-CHAIN

## Triglycerida saturata media

## DEFINITION

Mixture of triglycerides of saturated fatty acids, mainly of caprylic (octanoic) acid and of capric (decanoic) acid. Medium-chain triglycerides are obtained from the oil extracted from the hard, dried fraction of the endosperm of *Cocos nucifera* L. or from the dried endosperm of *Elaeis guineensis* Jacq.

**Content:** minimum 95.0 per cent of saturated fatty acids with 8 and 10 carbon atoms.

## CHARACTERS

**Appearance:** colourless or slightly yellowish, oily liquid.

**Solubility:** practically insoluble in water, miscible with alcohol, with methylene chloride, with light petroleum and with fatty oils.

## IDENTIFICATION

First identification: B, C.

Second identification: A, D.

- A. Heat 3.0 g under a reflux condenser for 30 min with 50 ml of a mixture of equal volumes of *alcohol R* and 2 M *alcoholic potassium hydroxide R*. Reserve 10 ml of the mixture for identification test D. To 40 ml of the mixture add 30 ml of *water R*, evaporate the alcohol and acidify the hot solution with 25 ml of *dilute hydrochloric acid R*. After cooling, shake with 50 ml of *peroxide-free ether R*. Wash the ether layer with 3 quantities, each of 10 ml, of *sodium chloride solution R*, dry over *anhydrous sodium sulphate R* and filter. Evaporate the ether and determine the acid value (2.5.1) of the residue, using 0.300 g. The acid value is 350 to 390.
- B. It complies with the test for saponification value (see Tests).
- C. It complies with the test for composition of fatty acids (see Tests).
- D. Evaporate 10 ml of the alcoholic mixture obtained in identification test A to dryness on a water-bath. Transfer the residue into a test-tube, add 0.3 ml of *sulphuric acid R* and close the test-tube with a stopper through which a U-shaped glass tube is inserted. One end of the U-tube is dipped into 3 ml of a 10 g/l solution of *tryptophan R* in a mixture of equal volumes of *sulphuric acid R* and *water R*. Heat the test-tube in a silicone-oil bath at 180 °C for 10 min and collect the liberated fumes in the tryptophan reagent. Heat the tryptophan reagent on a water-bath for 1 min. A violet colour develops.

## TESTS

**Appearance.** The substance to be examined is clear (2.2.1) and not more intensely coloured than reference solution Y<sub>3</sub> (2.2.2, Method I).

**Alkaline impurities.** Dissolve 2.00 g in a mixture of 1.5 ml of *alcohol R* and 3.0 ml of *ether R*. Add 0.05 ml of *bromophenol blue solution R*. Not more than 0.15 ml of 0.01 M *hydrochloric acid* is required to change the colour of the indicator to yellow.

**Relative density** (2.2.5): 0.93 to 0.96.

**Refractive index** (2.2.6): 1.440 to 1.452.

**Viscosity** (2.2.9): 25 mPas to 33 mPas.

**Acid value** (2.5.1): maximum 0.2.

**Hydroxyl value** (2.5.3, Method A): maximum 10.

**Iodine value** (2.5.4): maximum 1.0.

**Peroxide value** (2.5.5, Method A): maximum 1.0.

**Saponification value** (2.5.6): 310 to 360.

**Unsaponifiable matter** (2.5.7): maximum 0.5 per cent, determined on 5.0 g.

**Composition of fatty acids.** Gas chromatography (2.4.22, Method C).

Column:

- *material*: fused silica,
- *size*:  $l = 30$  m,  $\varnothing = 0.32$  mm,
- *stationary phase*: *macrogol 20 000 R* (film thickness 0.5 µm),

*Carrier gas*: helium for chromatography R.

*Flow rate*: 1.3 ml/min.

*Temperature*:

	Time (min)	Temperature (°C)
Column	0 - 1	70
	1 - 35	70 → 240
	35 - 50	240
Injection port		250
Detector		250

*Detection*: flame ionisation.

*Split ratio*: 1:100.

*Composition of the fatty acid fraction of the substance*:

- *caproic acid*: maximum 2.0 per cent,
- *caprylic acid*: 50.0 per cent to 80.0 per cent,
- *capric acid*: 20.0 per cent to 50.0 per cent,
- *lauric acid*: maximum 3.0 per cent,
- *myristic acid*: maximum 1.0 per cent.

**Chromium**: maximum 0.05 ppm, if intended for use in parenteral nutrition.

Atomic absorption spectrometry (2.2.23, Method II).

*Test solution.* Dissolve 2.0 g of the substance to be examined in *methyl isobutyl ketone R3* and dilute to 10.0 ml with the same solvent.

*Solution A.* Dilute 0.100 ml of *chromium liposoluble standard solution (1000 ppm Cr) R* to 10.0 ml with *methyl isobutyl ketone R3*.

*Stock solution.* Dilute 0.100 ml of solution A to 10.0 ml with *methyl isobutyl ketone R3*.

*Reference solutions.* Prepare 3 reference solutions by dissolving for each 2.0 g of the substance to be examined in the minimum volume of *methyl isobutyl ketone R3*, adding 0.5 ml, 1.0 ml and 2.0 ml, respectively, of stock solution and diluting to 10.0 ml with *methyl isobutyl ketone R3*.

*Source*: chromium hollow-cathode lamp.

*Wavelength*: 357.8 nm.

*Atomic generator*: graphite furnace.

*Carrier gas*: argon R.

**Copper**: maximum 0.1 ppm, if intended for use in parenteral nutrition.

Atomic absorption spectrometry (2.2.23, Method II).

*Test solution.* Dissolve 2.0 g of the substance to be examined in *methyl isobutyl ketone R3* and dilute to 10.0 ml with the same solvent.

*Solution A.* Dilute 0.100 ml of *copper liposoluble standard solution (1000 ppm Cu) R* to 10.0 ml with *methyl isobutyl ketone R3*.

*Stock solution.* Dilute 0.100 ml of solution A to 10.0 ml with *methyl isobutyl ketone R3*.

*Reference solutions.* Prepare 3 reference solutions by dissolving for each 2.0 g of the substance to be examined in the minimum volume of *methyl isobutyl ketone R3*, adding 1.0 ml, 2.0 ml and 4.0 ml, respectively, of stock solution and diluting to 10.0 ml with *methyl isobutyl ketone R3*.

*Source*: copper hollow-cathode lamp.

*Wavelength*: 324.7 nm.

*Atomic generator*: graphite furnace.

*Carrier gas*: argon R.

**Lead:** maximum 0.1 ppm, if intended for use in parenteral nutrition.

Atomic absorption spectrometry (2.2.23, Method II).

**Test solution.** Dissolve 2.0 g of the substance to be examined in *methyl isobutyl ketone R3* and dilute to 10.0 ml with the same solvent.

**Solution A.** Dilute 0.100 ml of *lead liposoluble standard solution (1000 ppm Pb) R* to 10.0 ml with *methyl isobutyl ketone R3*.

**Stock solution.** Dilute 0.100 ml of solution A to 10.0 ml with *methyl isobutyl ketone R3*.

**Reference solutions.** Prepare 3 reference solutions by dissolving for each 2.0 g of the substance to be examined in the minimum volume of *methyl isobutyl ketone R3*, adding 1.0 ml, 2.0 ml and 4.0 ml, respectively, of stock solution and diluting to 10.0 ml with *methyl isobutyl ketone R3*.

**Source:** lead hollow-cathode lamp.

**Wavelength:** 283.3 nm.

**Atomic generator:** graphite furnace coated inside with palladium carbide; calcination is carried out in the presence of oxygen at a temperature below 800 °C.

**Carrier gas:** argon R.

**Nickel:** maximum 0.2 ppm, if intended for use in parenteral nutrition.

Atomic absorption spectrometry (2.2.23, Method II).

**Test solution.** Dissolve 2.0 g of the substance to be examined in *methyl isobutyl ketone R3* and dilute to 10.0 ml with the same solvent.

**Solution A.** Dilute 0.100 ml of *nickel liposoluble standard solution (1000 ppm Ni) R* to 10.0 ml with *methyl isobutyl ketone R3*.

**Stock solution.** Dilute 0.100 ml of solution A to 10.0 ml with *methyl isobutyl ketone R3*.

**Reference solutions.** Prepare 3 reference solutions by dissolving for each 2.0 g of the substance to be examined in the minimum volume of *methyl isobutyl ketone R3*, adding 1.0 ml, 2.0 ml and 4.0 ml, respectively, of stock solution and diluting to 10.0 ml with *methyl isobutyl ketone R3*.

**Source:** nickel hollow-cathode lamp.

**Wavelength:** 232 nm.

**Atomic generator:** graphite furnace.

**Carrier gas:** argon R.

**Tin:** maximum 0.1 ppm, if intended for use in parenteral nutrition.

Atomic absorption spectrometry (2.2.23, Method II).

**Test solution.** Dissolve 2.0 g of the substance to be examined in *methyl isobutyl ketone R3* and dilute to 10.0 ml with the same solvent.

**Solution A.** Dilute 0.100 ml of *tin liposoluble standard solution (1000 ppm Sn) R* to 10.0 ml with *methyl isobutyl ketone R3*.

**Stock solution.** Dilute 0.100 ml of solution A to 10.0 ml with *methyl isobutyl ketone R3*.

**Reference solutions.** Prepare 3 reference solutions by dissolving for each 2.0 g of the substance to be examined in the minimum volume of *methyl isobutyl ketone R3*, adding 1.0 ml, 2.0 ml and 4.0 ml, respectively, of stock solution and diluting to 10.0 ml with *methyl isobutyl ketone R3*.

**Source:** tin hollow-cathode lamp.

**Wavelength:** 286.3 nm.

**Atomic generator:** graphite furnace coated inside with palladium carbide.

**Carrier gas:** argon R.

**Heavy metals (2.4.8):** maximum 10 ppm, if intended for use other than parenteral nutrition.

2.0 g complies with limit test D. Prepare the standard using 2 ml of *lead standard solution (10 ppm Pb) R*.

**Water (2.5.12):** maximum 0.2 per cent, determined on 10.00 g.

**Total ash (2.4.16):** maximum 0.1 per cent, determined on 2.0 g.

#### STORAGE

In a well-filled container, protected from light.

#### LABELLING

The label states, where applicable, that the substance is intended for use in parenteral nutrition.

01/2008:2032

## TRIGLYCEROL DIISOSTEARATE

### Triglyceroli diisostearas

#### DEFINITION

Mixture of polyglycerol diesters of mainly isostearic acid, obtained by esterification of polyglycerol and isostearic acid. The polyglycerol consists mainly of triglycerol.

#### CHARACTERS

**Appearance:** clear, yellowish, viscous liquid.

**Solubility:** practically insoluble in water, miscible with ethanol (96 per cent) and with fatty oils.

#### IDENTIFICATION

A. Infrared absorption spectrophotometry (2.2.24).

*Preparation:* film between 2 plates of *sodium chloride R*.

*Comparison:* *triglycerol diisostearate CRS*.

B. Composition of fatty acids (see Tests).

#### TESTS

**Appearance of solution.** The solution is not more intensely coloured than reference solution BY<sub>3</sub> (2.2.2, Method I).

Mix 10 ml with 10 ml of *ethanol (96 per cent) R*.

**Acid value (2.5.1):** maximum 3.0, determined on 1.0 g.

**Hydroxyl value (2.5.3, Method A):** 180 to 230, determined on 0.25 g.

**Iodine value (2.5.4, Method B):** maximum 3.0.

**Peroxide value (2.5.5, Method B):** maximum 6.0.

**Saponification value (2.5.6):** 128 to 160.

**Composition of fatty acids (2.4.22, Method B).** Use the mixture of calibrating substances in Table 2.4.22-1.

*Composition of the fatty-acid fraction of the substance:*

- *sum of the contents of the fatty acids eluting between palmitic acid and stearic acid:* minimum 60.0 per cent;
- *sum of the contents of myristic acid, palmitic acid and stearic acid:* maximum 11.0 per cent.

**Water (2.5.12):** maximum 0.5 per cent, determined on 2.00 g.